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of matter. Whether your reading light comes from a fluorescent lamp or the sun, the light's source is plasma. Although plasma is still somewhat mysterious to humans, the plasma state of matter is the most common state in the universe. In this chapter, your exploration of the states of matter will go far beyond everyday,

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142 Guided Reading and Study Workbook CHAPTER 13, States of Matter (continued) SECTION 13.3 THE NATURE OF SOLIDS (pages 396–399) This section describes the highly organized structures of solids, distinguishes between a crystal lattice and a unit cell, and explains how allotropes of an element differ. A Model for Solids (page 396) 1.

## **Name Date Class STATES OF MATTER 13**

Section 13.1 continued CHAPTER 13 STUDY GUIDE FOR CONTENT MASTERY Vacuum Atmospheric pressure Pressure exerted by mercury column 760 mm Name Date Class Study Guide for Content Mastery Chemistry: Matter and Change Chapter 13 73 States of Matter Section 13.1 Gases In your textbook, read about the kinetic-molecular theory. Complete each statement. 1.

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Chapter 13 states of matter. STUDY. PLAY. Kinetic-molecular theory. The kinetic theory of gases describes a gas as a large number of small particles (atoms or molecules), all of which are in constant, random motion. The rapidly moving particles

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constantly collide with each other and with the walls of the container.

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Temperature is best defined as: the average kinetic energy in a sample of matter 7. The state of a substance at a particular temperature depends mostly on: the strength of its intermolecular forces 8. List the characteristics of substances

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with strong intermolecular forces and weak intermolecular forces. ... Chapter 13 Study Guide Answers 9 ...

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