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14.2 Brønsted-Lowry
Acids and Bases 447
14.2 Brønsted-Lowry
Acids and Bases

LEARNING GOAL

Identify conjugate
acid-base pairs for
Brønsted-Lowry acids
and bases. In 1923, J.
N. Brønsted in
Denmark and T. M.
Lowry in Great Britain
expanded the defini-
tion of acids and bases

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to include bases that
do not contain OH⁻ ions.

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Acids, Bases, and
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A 1. A 2. C 3. D 4. A 5.

A smaller than the
particles in a 6. A 7. D

8. A 9. A 10. A 11.

colloid 12. separate the
ions from each other.

dissolve 13. corrosive

14. hydrochloric 15.

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base 16. F, water and a
salt 17. F, temperature
18. T 19. F, bitter 20. F
...

Answers Acids, Bases, and Solutions

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particles in a 6. A 7. D
8. A 9. A 10. A 11.
colloid 12. separate the

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14. hydrochloric 15.
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salt 17. F, temperature
18. T 19 ...

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An acid is any
compound that forms
 H^+ ions in solution.

Acidic solutions have

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pH values below 7. A
base is a compound
that forms OH^- ions in
solution. Basic, or

alkaline, solutions have
pH values above 7.

Buffers are weak acids
or bases that can react
with strong acids or
bases to prevent
sudden changes in pH.
The Water Molecule

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Acids, Bases, and
Solutions ANSWER KEY

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Acids, Bases, and
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Acids and Bases
Review and Reinforce

1. Sour 2. Bitter 3.

Corrosive to
magnesium, zinc, and
iron; eats them away
and produces bubbles
of hydrogen gas 4.

Doesn't react with
metals 5. Produces
carbon dioxide 6.

Doesn't react with
carbonates 7. Red 8.
Blue 9.

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**Acids, Bases, and
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Chapter 8 Solutions,
Acids & Bases

Dissolving:

Dissociation,

Dispersion or Ionization

Substances can

dissolve in water by

three ways: 1.

dissociation 2.

dispersion 3. ionization

Dissociation is the

process by which an

ionic compound

separates into ions Ex:

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NaCl Dispersion is the process by which a compound breaks into small pieces that spread throughout the water Ex: sugar dissolves in water ...

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Solutions for class 10
chapter 2 Acids, bases
and salt The salts of
strong acids and weak
bases give acidic
solution having pH less
than 7. Example, ...

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Chapter 7 Acids And
Bases | pdf Book ...

This section describes
the general properties
of acids and bases.

Reading Strategy
(page 240) Using Prior
Knowledge Before you
read, write your
definition of each

Chapter 8:
Page 20/26

Where To Download Acids Bases And **Solutions, Acids, and Bases**

You are given three clear, colorless, unknown solutions, and red and blue litmus papers. Describe the steps you might take to correctly identify each solution as acidic, basic, or neutral. 29. Describe how solutions of acids and bases differ in terms of particles. How do the differences relate to the properties of acids

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and bases? 30.

Acids, Bases, and Solutions

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Chemistry ...

Worked solutions: Chapter 7 Acids and bases

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Concepts in Action 1

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Physical Science

Concepts in Action.

Chapter 8 ; Solutions,
Acids Bases; 2 8.1

Formation of Solutions.

Objectives ; 1. Describe

how a substance can
dissolve in water by
dissociation, dispersion
or ionization ; 2.

Describe how physical
properties of a solution

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