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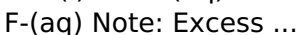
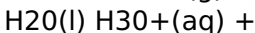
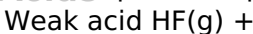
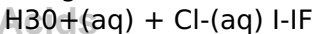
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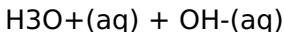
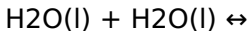
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strong acids include
HCl, HBr, HI, HClO₄,
H₂SO₄, and HNO₃ -
molecules of strong
acids completely ionize
in solution to produce
hydronium ions; 100%
of molecules of strong
acid are ionized in a

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solution (assuming concentration is not extremely high) - As such, concentration of H_3O^+ in strong acid solutions are equal to initial concentration of strong acid pH of strong acid solution is easily calculated - Common.

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solution and weak
acids are unlike the
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cannot be dissociated
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