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photon having wavelength 650 nm? Repeat for an electron
having the same wavelength and only kinetic energy. Solution:
For the photon, $p = hc/E$; $E = hc/\lambda = hc/650 \times 10^{-9} \dots$

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Required Textbook: " Fundamentals of Nanoelectronics" by George W. Hanson, Pearson/Prentice Hall (2008), ISBN 978-0131957084. RECOMMENDED READINGS: 1. "Mesoscopic Electronics in Solid State Nanostructures" by Thomas Heinzl. 2. "Nanoelectronics and Information Technology", 2 nd Ed. by Rainer Waser (Ed.) 3.

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