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4 Problems and Solutions in Atomic, Nuclear and Particle Physics Solution:

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The Bohr radius of the hydrogen atom and the Compton wavelength of an electron are given by $a_0 = \frac{h^2}{m_e e^2}$ and $\lambda_c = \frac{h}{m_e c}$ respectively. Hence $\frac{\lambda_c}{a_0} = \frac{1}{2\pi} \left(\frac{e^2}{c} \right) = \frac{1}{137} \approx 0.0073$, where $\frac{e^2}{c}$ is the fine-structure constant. Hence the answer is (a). 1004
Estimate the electric field needed to pull an electron out of an atom in

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Modern Atomic Physics | Optics & Photonics News

$\frac{1}{f(v, 1)} = -0.000814/6.967 = -0.0001168$ The second approximation, therefore, gives $a = -2.1039291$. The third and higher approximations can be made in this fashion. The first

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approximation will be usually good enough in practice. $1.34y(x) = x^2 \exp(-x^2)$ (1) Turning points are determined from the location of maxima and minima.

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